

UNIVERSITY OF CALCUTTA
VIDYASAGAR COLLEGE FOR WOMEN CENTRE
B.Sc. SEM 1 Chemistry General Practical Examination 2021
Subject: CEMG- CC1/GE1
Full Marks: 30

Estimate the strength of oxalic acid in g/L in a given solution by titrating against the KMnO_4 solution

Given

1. Standardization of $\sim(\text{N}/20)$ KMnO_4 solution by standard $(\text{N}/20)$ Oxalic acid solution

V_{Oxalic} = Volume of Oxalic acid solution pipetted out = 25 ml

S_{Oxalic} = Strength of Oxalic acid solution = $0.995(\text{N}/20)$

V_{KMnO_4} = Volume of KMnO_4 solution required for titration = 22.8 ml

2. Neutralization of supplied oxalic acid solution by standard KMnO_4 solution

V_{Oxalic} = Volume of supplied Oxalic acid solution pipetted out = 25 ml

V_{KMnO_4} = Volume of KMnO_4 required for neutralization = 26.3 ml

Write the experimental result with respect to

- a) Theory
- b) Tabular form of experimental data
- c) Calculation and result